## CCE RR <br> UNREVISED FULL SYLLABUS

 KARNATAKA SCHOOL EXAMINATION AND ASSESSMENT BOARD, MALLESHWARAM, BENGALURU - 560003

S. S. L. C. EXAMINATION, JUNE - 2023

యూదర లుత్తరగగఆక
MODEL ANSWERS
దినృంళ : 13. 06. 2023 ]
Date: 13.06. 2023 ]

Code no. : 83-E (Phy)

## 

## Subject : SCIENCE

(భౌత విజ్ఞాన, రనాయిన విజ్ఞాన ముత్తు జిల విజ్ఞాన / Physics, Chemistry \& Biology )
( ష్లుసరాజతికత లాలా అభ్యథీ / Regular Repeater )
( భౌత విజ్ఞాన / Physics )
( ఇంగ్లిషో యూధ్యయు / English Medium )
[ Max. Marks : 80

PART - A
(Physics)

| Qn. <br> Nos. | Value Points | Total |  |
| :--- | :--- | :---: | :---: |
| I. | Multiple choice questions : | $\mathbf{4} \times \mathbf{1}=\mathbf{4}$ |  |
| 1. | A device that converts electrical energy into mechanical  <br> energy is  <br> (A) Electric generator (B) Electric motor <br> (C) Galvanometer (D) Voltmeter. <br> Ans. :  <br> (B) Electric motor  |  |  |


| Qn. Nos. | Value Points | Total |
| :---: | :---: | :---: |
| 2. | A mirror forms an erect and enlarged image of an object. Then the type of the mirror and the nature of the image respectively are <br> (A) convex mirror and virtual image <br> (B) concave mirror and real image <br> (C) plane mirror and real image <br> (D) concave mirror and virtual image. <br> Ans. : <br> (D) concave mirror and virtual image | 1 |
| 3. | The power plant that generates electricity without using the turbines is <br> (A) Thermal power plant <br> (B) Hydro power plant <br> (C) Solar power plant <br> (D) Nuclear power plant. <br> Ans. : <br> (C) Solar power plant | 1 |
| 4. | Imagine, you are holding a straight current carrying conductor as per the right hand thumb rule. If the thumb is upward, then the direction of the field lines of the magnetic field is <br> (A) downward <br> (B) upward <br> (C) anti-clockwise <br> (D) clockwise. <br> Ans. : <br> (C) anti-clockwise | 1 |
| II. | Answer the following questions : $2 \times 1=2$ <br> Draw the symbol diagram of rheostat used in electric circuit. <br> Ans. : | 1 |




| Value Points |  |
| :--- | :--- |
| $R_{1}=\rho \frac{l}{A}=4 \Omega$ | $\frac{1}{2}$ |
| Now for second wire |  |
| $R_{2}=\rho \frac{\frac{l}{2}}{2 A}$ | $\frac{1}{2}$ |
|  | $=\frac{1}{4} \cdot \rho \frac{l}{A}$ |
| $R_{2}$ | $=\frac{1}{4} \cdot R_{1}$ |

$\therefore \quad$ The resistance of the another wire is

$$
\frac{1}{4} \cdot \not 4=1 \Omega
$$

Answer the following questions :
$\mathbf{3 \times 3}=\mathbf{9}$
9. What is meant by the 'aperture' of a spherical mirror ? Mention the four uses of a concave mirror.

## OR

a) What is meant by the power of a lens ? Write the formula used to find the power of a lens. What is the SI unit of power of a lens ?
b) If the focal lengths of two lenses $A$ and $B$ are +0.50 m and -0.40 m respectively. Mention the types of these lenses in the same order.

Ans. :
$\star$ The diameter of the reflecting surface of spherical mirror.
$\star$ Used in torches, search-lights and vehicle head lights to get parallel beam of light
^ as a shaving mirror
$\star$ by dentists to see large images of the teeth
$\star$ in solar furnaces to concentrate sunlight

| $\underset{\substack{\text { Qn. } \\ \text { Nos. }}}{\text {. }}$ | Value Points | Total |
| :---: | :---: | :---: |
| a) | $\star \quad$ The degree of convergence or divergence of light rays is the power of a lens <br> * $\quad P=\frac{1}{f}$ |  |
| b) | $\begin{array}{lll} \star & \text { SI unit of power of a lens is 'dioptre'. OR 'D' } & \frac{1}{2} \\ \star & +0.50 \mathrm{~m} \rightarrow \text { Convex lens } & \frac{1}{2} \end{array}$ |  |
|  | $\star-0.40 \mathrm{~m} \rightarrow \text { Concave lens } \quad \frac{1}{2}$ | 3 |
| 10. | Observe the given diagram : |  |
|  | Explain the experiment related to this diagram. What conclusions can be drawn from this experiment? <br> Ans. : |  |
|  | $\star$ Take two different coils of copper wire say 100 and 50 turns respectively. Insert them over a non-conducting cylindrical roll. |  |
|  | $\star$ Connect the Coil-1 in series with a battery and plug key, Coil-2 with galvanometer <br> $\star \quad$ When the key is plugged in, needle of the galvanometer deflects and returns to zero. This indicates the presence of current in the Coil-2 |  |

$\star$ Disconnect Coil-1 from battery. Needle of the galvanometer deflects in the opposite direction and returns to zero. This indicates the opposite direction of the current.

Conclusions:
$\star$ Changing electric current in Coil-1 induces current in Coil-2. This is electromagnetic induction. $\frac{1}{2}$
$\star \quad$ This is due to the change in the magnetic field. $\quad \frac{1}{2}$

Draw the ray diagram for the image formation by a convex lens, when the object is placed at $2 F_{1}$. With the help of the ray diagram mention the position and the nature of the image formed.
[ $F_{1}$ : Principal focus of the lens ]

## OR

Draw the ray diagram for the image formation in a convex lens when the object is placed beyond $2 F_{1}$. With the help of the ray diagram mention the position and the nature of the image formed.
[ $F_{1}$ : Principal focus of the lens ]

Ans. :

| Qn. Nos. | Value Points | Total |
| :---: | :---: | :---: |
|  |  |  |
|  | $\star$ Ray diagram 2 <br> $\star$ Position of the image : at $2 F_{2}$ $\frac{1}{2}$ <br> $\star$ Nature of the image : Real and inverted $\frac{1}{2}$ | 3 |
|  |  |  |
|  | * Ray diagram <br> $\star \quad$ Position of the image : between $F_{2}$ and $2 F_{2}$ <br> $\star \quad$ Nature of the image : Real and inverted. | 3 |
| v. | Answer the following question : $1 \times 4=4$ |  |
| 12. | a) A bread-toaster rated 350 W is used for 15 hours a day. An electric iron box rated 250 W is used for 5 hours a day. Calculate the cost of using these appliances for 30 days, if the cost of 1 kWh is Rs. 4. |  |


13.
a) How does the lens of human eye accommodate to see the nearby objects and the distant objects ? Explain.
b) Explain the formation of rainbow in the nature.

Ans. :


* When the ciliary muscles are relaxed the eye lens becomes thin
$\star$ This increases its focal length
$\star$ and the distant objects can be seen clearly
$\star \quad$ When the ciliary muscles contract the eye lens becomes thick
$\star$ This decreases its focal length
$\star$ and the nearby objects can be seen clearly.
b) $\star$ The water droplets act like small prisms
$\star \quad$ They refract and disperse the incident sunlight They refract and disperse the incident sunlight
VI.

Answer the following question : $1 \times 5=5$
The value of the current increases.
Parallel connection

| Qn. <br> Nos. | Value Points | Total |  |
| :---: | :---: | :---: | :---: |
|  | $\star$ | Then reflect internally | $\frac{1}{2}$ |
|  | $\star$ | Finally refract again while coming out of water droplets. |  |
|  | Due to the dispersion of light in this manner the |  |  |
|  | rainbow is formed. | $\frac{1}{2}$ | 5 |

## CCE RR <br> UNREVISED FULL SYLLABUS

## A

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S. S. L. C. EXAMINATION, JUNE - 2023

MODEL ANSWERS

దనాంళ : 13. 06. 2023 ]
Date: 13.06.2023]
 Code no. : 83-E (Chem.)

## ఎిజయ : ఎిజ్ఞాన

## Subject : SCIENCE


( 山్లుNరల山తిఁత్ర లలలల అభ్యథీ / Regular Repeater )
( రనౌయన విజ్లానె / Chemistry )
( ఇంగ్లిఱో యూధ్యయు / English Medium )

[ Max. Marks : 80

PART - B
(Chemistry )

| Qn. <br> Nos. | Value Points | Total |
| ---: | :--- | :---: |
| VII. | Multiple choice questions : | $\mathbf{2} \times \mathbf{1 = 2}$ |
| 14. | Mendeleev's periodic table is constructed on the basis of |  |
|  | (A) Atomic number |  |
| (B) |  |  |
|  | Electronic configuration of an atom |  |
|  | (C) Atomic size |  |
| (D) Atomic mass. |  |  |
| Ans. : |  |  |
|  | (D) Atomic mass | 1 |

RR-A (MA)-CHE


| Qn. <br> Nos. | Value Points | Total |
| ---: | :--- | :---: |
| 19. | Ionic compounds have high melting point and boiling point. <br> Why? |  |
|  | Ans. : | Considerable amount / more amount of energy is required <br> to break the strong inter ionic attraction between the <br> molecules. |

In a homologous series, the first member of hydrocarbon group has the molecular formula $\mathrm{CH}_{4}$. Then find out the molecular formula of the fourth member and write two types of structural formula of it.

Ans. :
$\star \quad \mathrm{C}_{1} \mathrm{H}_{4}$

$$
\begin{aligned}
& \mathrm{C}_{n} \mathrm{H}_{2 n+2} \\
& \mathrm{C}_{4} \mathrm{H}_{(2 \times 4)+2}
\end{aligned}
$$

$\frac{\mathrm{C}_{1} \mathrm{H}_{2}}{\mathrm{C}_{2} \mathrm{H}_{6}}$
OR
$\star \quad \frac{\mathrm{C}_{1} \mathrm{H}_{2}}{\mathrm{C}_{3} \mathrm{H}_{8}}$
$\mathrm{C}_{4} \mathrm{H}_{(8+2)}$
$\star \quad \frac{\mathrm{C}_{1} \mathrm{H}_{2}}{\mathrm{C}_{4} \mathrm{H}_{10}}$

$$
\begin{equation*}
\mathrm{C}_{4} \mathrm{H}_{10} \tag{1}
\end{equation*}
$$

Butane ( $\mathrm{C}_{4} \mathrm{H}_{10}$ ) structures



OR



$\frac{1}{2}$
2
RR-A (MA)-CHE
[ Turn over bronze and solder metal.

## OR

What are ores ? Name the respective methods used to convert sulphide and carbonate ores of metals into their oxides.

Ans. :
$\star$ An alloy is a homogenous mixture of two or more metals or metals and non-metals.
$\star$ Bronze - Copper and tin / Cu and $\mathrm{Sn} \quad \frac{1}{2}$
$\star \quad$ Solder metal — Lead and tin / Pb and $\mathrm{Sn} \quad \frac{1}{2}$

## OR

$\star$ Minerals contain a very high percentage of a particular metal and the metal can be profitably extracted from it.
$\star$ Metallic sulphide ore — Roasting $\quad \frac{1}{2}$
$\star$ Metallic carbonate ore - Calcination $\quad \frac{1}{2}$
Add same amount of barium chloride solution to a test tube containing 5 ml of sodium sulphate solution. Then
i) Which insoluble white precipitate is formed ?
ii) Name the ions responsible for the formation of white precipitate.
iii) Mention the type of chemical reaction that took place here.
Ans. :
i) $\mathrm{BaSO}_{4} /$ barium sulphate
ii) $\mathrm{SO}_{4}^{2-}-$ sulphate radical
$\mathrm{Ba}^{2+}-$ Barium ion
iii) Double displacement reaction / precipitation reaction.

24. The elements are arranged in the increasing order of their atomic masses in the below given table. Observe it and answer the following questions :

| Sa | Re | Ga | Ma | Pa | Dha | Ni |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| H | Li | Be | B | C | N | O | F | Na |

i) Name the elements that belong to the same group.
ii) State the law that helps to group these elements.
iii) Write two limitations of the same law.

Ans. :
i) H and F $\quad \frac{1}{2}$

Li and $\mathrm{Na} \quad \frac{1}{2}$
RR-A (MA)-CHE
[ Turn over
ii) Newlands' law of octaves.

When the elements arranged in the order of increasing atomic masses, every eighth element had properties similar to that of first.
iii) Limitations:
$\star$ Applicable only up to calcium

* Wrong guess made such as 'no more elements would be discovered in future'.
$\star \quad$ Adjusted two unsimilar elements in the same slot
$\star$ With the discovery of noble gases the law of octaves become irrelevant.

$$
\begin{equation*}
\text { ( Any two points ) } \quad \frac{1}{2}+\frac{1}{2} \tag{3}
\end{equation*}
$$

a) Identify unsaturated hydrocarbons in the following carbon compounds and write their structural formula.

$$
\mathrm{C}_{6} \mathrm{H}_{6}, \quad \mathrm{C}_{5} \mathrm{H}_{12}, \quad \mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}, \quad \mathrm{C}_{2} \mathrm{H}_{2}
$$

b) Write the difference between esterification and saponification.

## OR

a) Write electron dot structure of oxygen molecule.
b) Carbon atom does not form $\mathrm{C}^{4-}$ anion and $\mathrm{C}^{4+}$ cation. Why ?

Ans. :
a)

Unsaturated hydrocarbons | Structural formula | $\frac{1}{2}+\frac{1}{2}$ |  |
| :---: | :---: | :---: |
| $\mathrm{C}_{2} \mathrm{H}_{2}$ | $\mathrm{H}-\mathrm{C} \equiv \mathrm{C}-\mathrm{H}$ | $\frac{1}{2}+\frac{1}{2}$ |

| Qn. Nos. | Value Points | Total |
| :---: | :---: | :---: |
| b) | Esterification : Reaction between an acid and an alcohol to produce esters. $\frac{1}{2}$ <br> Saponification : Reaction between an alkaline base and long chain carboxylic acid to produce soaps [ or sodium / potassium salts of long chain carboxylic acid ] $\frac{1}{2}$ | 3 |
| a) | OR <br> $\mathrm{O}=\mathrm{O}$ |  |
| b) | $\star \mathrm{C}^{4-}$ anion does not form because difficult for the nucleus with six protons to hold on ten electrons. 1 $\mathrm{C}^{4+}$ cation does not form because require large amount of energy to remove four electrons leaving behind a carbon with six proton in its nucleus holding on just two electrons. | 3 |
| XI. | Answer the following question : $1 \times 4=4$ |  |
| 26. | a) Explain the manufacturing of bleaching powder. Write any two uses of it. <br> b) A strong solution of sodium hydroxide is added to the strong solution of hydrochloric acid. What is the nature of the salt solution formed here ? Write a balanced chemical equation for this reaction. |  |


| Qn. Nos. | Value Points | Total |
| :---: | :---: | :---: |
|  | Ans. : |  |
| a) | Bleaching powder is produced by the action of chlorine on dry slaked lime. |  |
|  | OR |  |
|  | $\mathrm{Ca}(\mathrm{OH})_{2}+\mathrm{Cl}_{2} \rightarrow \mathrm{CaOCl}_{2}+\mathrm{H}_{2} \mathrm{O} \longrightarrow 1$ |  |
|  | Uses : |  |
|  | $\star$ For bleaching cotton and linen in the textile industry, wood pulp in paper factories. |  |
|  | $\star$ For bleaching washed clothes in laundry |  |
|  | $\star$ As an oxidising agent in chemical industry |  |
|  | $\star$ to make drinking water free from germs. |  |
|  | (Any two ) $\frac{1}{2}+\frac{1}{2}$ |  |
| b) | $\star$ The salt solution is a neutral solution. 1 |  |
|  | $\star \mathrm{NaOH}+\mathrm{HCl} \rightarrow \mathrm{NaCl}+\mathrm{H}_{2} \mathrm{O} . \quad 1$ | 4 |

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S. S. L. C. EXAMINATION, JUNE - 2023

యూదరి లుత్తరగగళక
MODEL ANSWERS

దినృంళ : 13. 06. 2023 ]
Date: 13.06.2023]

Code No.: 83-E (Bio)

## 

## Subject : SCIENCE

(భౌత విజ్ఞాస, రనాయున విజ్ఞాన ముత్తు జియ విజ్ఞాన / Physics, Chemistry \& Biology ) ( ష్లుసరాజతికత లాలా అభ్యథీ / Regular Repeater )
( జిలప విజ్ఞూన / Biology )
( ఇంగ్లిఱఱ ふోధ్యయు / English Medium )

[ Max. Marks : 80

PART - C
( Biology )

| Qn. <br> Nos. | Value Points | Total |  |
| ---: | :--- | ---: | :---: |
| XII. | Multiple choice questions : | $\mathbf{2} \times \mathbf{1}=\mathbf{2}$ |  |
| 27. | Producers of aquatic eco-system are |  |  |
|  | (A) algae | (B) small fishes |  |
|  | (C) larvae | protozoa. |  |
| Ans. : |  |  |  |
| (A) algae |  |  |  |


| Qn. |
| ---: | :--- | :--- |
| Nos. | Biological process that has been shown in the diagram is


| $\begin{aligned} & \text { Qn. } \\ & \text { Nos. } \end{aligned}$ | Value Points | Total |
| :---: | :---: | :---: |
| $\begin{array}{r} \text { XIV. } \\ 31 . \end{array}$ | Answer the following questions : <br> What needs of the local people are fulfilled by the forest ? <br> Ans. : <br> Local people obtain : <br> $\star$ Large quantities of firewood, small timber and grass. <br> $\star$ Bamboo to make slats for huts and baskets for collecting and storing food materials. <br> ^ Essential materials to prepare the implements for agriculture, fishing and hunting. <br> $\star$ Fruits, nuts and medicines. <br> $\star \quad$ Grazing area for their cattle. |  |

Draw the diagram showing the structure of nephron and label 'glomerulus'.

Ans. :


Structure of nephron

Figure -
Part -
$1 \frac{1}{2}$
$\frac{1}{2}$

| Qn. Nos. | Value Points | Total |
| :---: | :---: | :---: |
| 33. | Student ' $A$ ' tells to Student ' $B$ ' that the wing of bird and arm of human are analogous organs. Student ' $B$ ' replies both of them are homologous organs. Whose answer is correct ? Justify your answer with suitable reasons. <br> Ans. : <br> Student B's answer is correct. <br> Because, <br> $\begin{array}{lll}\star & \text { they might be evolved from a common ancestor } & \frac{1}{2} \\ \star & \text { the basic structure of wing and arm is similar } & \frac{1}{2} \\ \star & \text { they perform different functions } & \frac{1}{2}\end{array}$ | 2 |
| XV. <br> 34. | Answer the following questions: $3 \times 3=9$ <br> Draw the diagram showing the structure of human brain and label the following parts : <br> i) Mid-brain <br> ii) Pons <br> Ans. : |  |
|  |  |  |
|  | Diagram - 2 <br> Part $-\frac{1}{2}+\frac{1}{2}$ | 3 |
| 35. | Round, green colour seeds producing pea plant ( $R R y y$ ) are crossed with wrinkled, yellow colour seeds producing pea plant ( $r r Y Y$ ). Show the result of $F_{2}$ generation with the |  |





| Qn. Nos. | Value Points | Total |
| :---: | :---: | :---: |
| b) | $\star$ In plants the large intercellular spaces and all the cells are oftenly in contact with air, due to this $\mathrm{CO}_{2}$ and oxygen are exchanged by diffusion here. This means $\frac{1}{2}$ <br> ^ Gases can go into cells and away from them and out into the air / atmosphere. <br> OR | 4 |
| a) | Human heart <br> $\star$ Has different chambers <br> $\star$ The valves present in between the chambers prevent backward flow of blood <br> * Separated by dividing wall septum <br> ^ Septum is responsible for creating separate pathways to transport oxygenated and deoxygenerated blood. $\frac{1}{2}$ |  |
| b) | Absorbed by finger like projections Villi present in small intestine <br> $\star$ Blood plasma - transports food, carbon dioxide and nitrogne wastes <br> $\star$ RBC — Carries oxygen <br> $\star$ Many other substances like salts are also transported by blood. | 4 |

